

**ADMISSION & EXAMINATION  
BYE-LAWS**

**FOR**

**MASTER OF PHARMACY  
IN  
PHARMACEUTICAL ANALYSIS**

**Program Code: MPA**



**SCHOOL OF PHARMACEUTICAL EDUCATION AND RESEARCH  
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## BOS MEETING DETAILS

- **Approval date of the BOS/School Board meeting for the present syllabus:**

<b>Name of the program</b>	<b>Department</b>	<b>Board of School (BOS) Approval Date</b>
M. Pharm	Pharmaceutical Analysis	21.04.2017

- **Approval date of the Academic Council meeting for the present syllabus**

<b>Name of the program</b>	<b>Program Code</b>	<b>Dates of Revision</b>
M. Pharm	MPA	31.05.2017

## VISION AND MISSION STATEMENTS

**Vision Statement:** To be the exemplary learning centre of excellence in pursuit of newer heights in higher education, research and innovation in Pharmaceutical Analysis.

### **Mission Statements:**

**MS-1:** To impart professional education through structured program in tune with the needs of the industry.

**MS-2:** To promote development of analytical skills required for analytical method development

**MS-3:** To provide training and practice for spectral analysis of organic compounds

**MS-4:** To develop collaborations at national and international levels to facilitate Industry-Academia relationship

## PROGRAM EDUCATIONAL OBJECTIVES (PEOs)

**After completing M. Pharm course in Pharmaceutical Analysis, the student should be able to:**

**PEO-1:** Apply the acquired knowledge and skills for the analytical / bio analytical method development and validation.

**PEO-2:** Acquire expertise in analysis and quality control of drugs, food products, pesticides in food, herbals and cosmetics.

**PEO-3:** Promote continuous update of knowledge and skills in the field of pharmaceutical analysis with proficiency in advanced pharmaceutical and instruments analysis for employment opportunities in various organisations.

**PEO-4:** Apply analytical knowledge and skills for interpretation of NMR, Mass and IR spectra of various organic compounds.

**PEO-5:** Understand quality control and quality assurance aspects of pharmaceutical industries and scope of quality certifications and regulatory affairs applicable to pharmaceutical industries.

**PEO-6:** Develop leadership skills to be applied in R&D, production and other facets of the profession

### Mapping Program Educational Objectives (PEOs) with Mission Statements (MS)

	MS-1	MS-2	MS-3	MS-4
<b>PEO-1</b>	3	2	3	3
<b>PEO-2</b>	3	3	3	2
<b>PEO-3</b>	3	3	3	3
<b>PEO-4</b>	2	2	3	3
<b>PEO-5</b>	3	3	3	2
<b>PEO-6</b>	3	3	2	3

**Level of Mapping:** '3' is for 'high-level' mapping, 2 for 'Medium-level' mapping, 1 for 'Low-level' mapping.

### PROGRAM OUTCOME (PO`s)

**After completing M. Pharm in Pharmaceutical Analysis, the student should be able to:**

- PO-1 Pharmacy Knowledge:** Apply domain knowledge and skills associated with Pharmaceutical analysis for quality control and analysis of drugs and their formulations, finished food products, herbals and cosmetics as per their regulatory requirements.
- PO-2 Planning abilities:** Identify and formulate research problems relating to analytical /bioanalytical the method development; plan, execute, report and interpret results of an experiment effectively with time and resource management.
- PO-3 Problem analysis:** Demonstrate efficiency in problem solving skills, based on scientific and analytical approaches through various strategies in pharmaceutical analysis.
- PO-4 Modern tool usage:** Select and apply latest scientific methods, resources and computing tools to analyse drugs with thorough understanding of limitations.
- PO-5 Leadership skills:** Demonstrate capability to build and lead team that can help to achieve the goal, inspire and motivate team members to engage for achieving results and use leadership skills to lead the team for accomplishment of goals in a responsible way.
- PO-6 Professional identity:** Demonstrate attitude, values, knowledge and skills that are key to fulfil professional responsibility. Understand, analyse and communicate the value of their professional roles in society as scientists, analyst, health care professionals, and educators.

**PO-7 Pharmaceutical ethics:** Apply ethical principles while making decisions and take responsibility for outcomes associated with the decisions. Demonstrate behaviour that respects cultural and personal variability in values, communication and life styles.

**PO-8 Communication:** Communicate effectively with scientific community and with society at large, in writing and orally; express views, thoughts and ideas in a clear and concise manner. Inscribe scientific reports, compose effectual presentations and documentation and offer and obtain comprehensible instructions.

**PO-9 The Pharmacist and society:** Demonstrate responsible behaviour and ability to assess community health, safety and legal issues and consequent responsibilities relevant to professional practice.

**PO-10 Environment and sustainability:** Understand the effect of chemicals and solvents used in pharmaceutical analysis on environment and provide environment friendly solutions to reduce negative impact through sustainability.

**PO-11 Lifelong learning:** Promote lifelong learning activities through self-motivation focussed at personal and professional developments; to fulfil these needs attend and participate in scientific seminars / conferences / workshop on an ongoing basis.

### **PROGRAM SPECIFIC OUTCOME (PSOs)**

**After completing M. Pharm course in Pharmaceutical Analysis, the student should be able to:**

**PSO-1:** Acquire analytical skills required for analytical / bioanalytical methods development and validation.

**PSO-2:** Apply skills related to theoretical and practical aspects of advanced/ hyphenated instrumental techniques for identification, characterisation and quantification of drugs and organic compounds.

**PSO-3:** Acquire expertise in analysis and quality control of drugs and their formulations, finished food products, herbal and cosmetic products with focus on their regulatory requirements.

**PSO-4:** Apply the understanding in quality control and quality assurance aspects of pharmaceutical industries for quality certifications and regulatory affairs.

**Mapping of Program Outcomes (POs) and Program Specific Outcomes (PSOs) with  
Program Educational Objectives (PEOs)**

	<b>PEO-1</b>	<b>PEO-2</b>	<b>PEO-3</b>	<b>PEO-4</b>	<b>PEO-5</b>	<b>PEO-6</b>
<b>PO-1</b>	3	3	3	2	3	3
<b>PO-2</b>	2	2	3	3	2	3
<b>PO-3</b>	2	3	3	3	2	1
<b>PO-4</b>	3	3	3	2	1	2
<b>PO-5</b>	2	3	3	3	3	3
<b>PO-6</b>	2	3	3	2	3	3
<b>PO-7</b>	2	2	2	3	2	1
<b>PO-8</b>	2	2	3	3	3	3
<b>PO-9</b>	2	2	3	3	3	3
<b>PO-10</b>	3	2	3	3	2	1
<b>PO-11</b>	3	3	3	3	3	3
<b>PSO-1</b>	3	3	3	2	2	3
<b>PSO-2</b>	3	3	3	2	2	3
<b>PSO-3</b>	3	3	3	2	2	3

**Level of Mapping:** '3' is for 'high-level' mapping, 2 for 'Medium-level' mapping, 1 for 'Low-level' mapping.

## CONSOLIDATED SEMESTER WISE PROGRAMME DETAILS

**Tables-I: Schemes for internal assessments and end semester examinations semester wise**

### Semester I

Course code	Name of the course	Internal Assessment				End Semester Exams		Total Marks	Credit points
		Continuous mode	Sessional Exams		Total	Marks	Duration		
			Marks	Duration					
MPA101T	Modern Pharmaceutical Analytical Techniques	10	15	1 Hr	25	75	3 Hrs	100	4
MPA102T	Advanced Pharmaceutical Analysis	10	15	1 Hr	25	75	3 Hrs	100	4
MPA103T	Pharmaceutical Validation	10	15	1 Hr	25	75	3 Hrs	100	4
MPA104T	Food Analysis	10	15	1 Hr	25	75	3 Hrs	100	4
MPA105P	Pharmaceutical Analysis Practical I	20	30	6 Hrs	50	100	6 Hrs	150	6
-	Seminar/Assignment	-	-	-	-			100	4
<b>Total</b>								<b>650</b>	<b>26</b>

### Semester II

Course code	Name of the course	Internal Assessment				End Semester Exams		Total Marks	Credit points
		Continuous Mode	Sessional Exams		Total	Marks	Duration		
			Marks	Duration					
MPA201T	Advanced Instrumental Analysis	10	15	1 Hr	25	75	3 Hrs	100	4
MPA 202T	Modern Bio-Analytical Techniques	10	15	1 Hr	25	75	3 Hrs	100	4
MPA 203T	Quality Control and Quality Assurance	10	15	1 Hr	25	75	3 Hrs	100	4
MPA 204T	Herbal and Cosmetic Analysis	10	15	1 Hr	25	75	3 Hrs	100	4
MPA 205P	Pharmaceutical Analysis Practical II	20	30	6 Hrs	50	100	6 Hrs	150	6
-	Seminar/Assignment							100	4
<b>Total</b>								<b>650</b>	<b>26</b>



### Semester III

Course code	Name of the course	Internal Assessment				End Semester Exams		Total Marks	Credit points
		Continuous Mode	Sessional Exams		Total	Marks	Duration		
			Marks	Duration					
MRM 301T	Research Methodology and Biostatistics*	10	15	1 Hr	25	75	3 Hrs	100	4
-	Journal club	-	-	-	25		3 Hrs	25	1
-	Discussion / Presentation (Proposal Presentation)	-	-	-	50		3 Hrs	50	2
-	Research Work	-	-	-	-	350	1 Hrs	350	14
<b>Total</b>								<b>525</b>	<b>21</b>

\* Non-University Exam

### Semester IV

Course code	Name of the course	Internal Assessment				End Semester Exams		Total Marks	Credit points
		Continuous Mode	Sessional Exams		Total	Marks	Duration		
			Marks	Duration					
-	Journal club	-	-	-	25	-	-	25	1
-	Discussion / Presentation (Proposal Presentation)	-	-	-	75	-	-	75	16
-	Research work and Colloquium	-	-	-	-	400	1 Hr	400	3
<b>Total</b>								<b>500</b>	<b>20</b>

## **RULES AND REGULATIONS**

### **1. Short Title and Commencement**

These regulations shall be called as “The Revised Regulations for the Master of Pharmacy (M. Pharm.) Degree Program - Credit Based Semester System (CBSS) of the Pharmacy Council of India, New Delhi”. They shall come into effect from the Academic Year 2016-17. The regulations framed are subject to modifications from time to time by the authorities of the university.

### **2. Minimum qualification for admission**

A Pass in the following examinations

- a) B. Pharm Degree examination of an Indian university established by law in India from an institution approved by Pharmacy Council of India and has scored not less than 55 % of the maximum marks (aggregate of 4 years of B.Pharm.)
- b) Every student, selected for admission to post graduate pharmacy program in any PCI approved institution should have obtained registration with the State Pharmacy Council or should obtain the same within one month from the date of his/her admission, failing which the admission of the candidate shall be cancelled.

Note: It is mandatory to submit a migration certificate obtained from the respective university where the candidate had passed his/her qualifying degree (B. Pharm.)

### **3. Duration of the program**

The program of study for M. Pharm. shall extend over a period of four semesters (two academic years). The curricula and syllabi for the program shall be prescribed from time to time by Pharmacy Council of India, New Delhi.

### **4. Medium of instruction and examinations**

Medium of instruction and examination shall be in English.

### **5. Working days in each semester**

Each semester shall consist of not less than 100 working days. The odd semesters shall be conducted from the month of June/July to November/December and the even semesters shall be conducted from the month of December/January to May/June in every calendar year.

### **6. Attendance and progress**

A candidate is required to put in at least 80% attendance in individual courses considering theory and practical separately. The candidate shall complete the prescribed course satisfactorily to be eligible to appear for the respective examinations.

### **7. Program/Course credit structure**

As per the philosophy of Credit Based Semester System, certain quantum of academic work viz. theory classes, practical classes, seminars, assignments, etc. are measured in terms of credits. On satisfactory completion of the courses, a candidate earns credits. The amount of credit associated with a course is dependent upon the number of hours of instruction per week in that course. Similarly, the credit associated with any of the other academic, co/extracurricular activities is dependent upon the quantum of work expected to be put in for each of these activities per week/per activity.

## **8 Credit assignment**

### **Theory and Laboratory courses**

Courses are broadly classified as Theory and Practical. Theory courses consist of lecture (L) and Practical (P) courses consist of hours spent in the laboratory. Credits (C) for a course is dependent on the number of hours of instruction per week in that course and is obtained by using a multiplier of one (1) for lecture and a multiplier of half (1/2) for practical (laboratory) hours. Thus, for example, a theory course having four lectures per week throughout the semester carries a credit of 4. Similarly, a practical having four laboratory hours per week throughout semester carries a credit of 2. The contact hours of seminars, assignments and research work shall be treated as that of practical courses for the purpose of calculating credits. i.e., the contact hours shall be multiplied by 1/2. Similarly, the contact hours of journal club, research work presentations and discussions with the supervisor shall be considered as theory course and multiplied by 1.

## **9. Minimum credit requirements**

The minimum credit points required for the award of M. Pharm. degree is 95. However based on the credit points earned by the students under the head of co-curricular activities, a student shall earn a maximum of 100 credit points. These credits are divided into Theory courses, Practical, Seminars, Assignments, Research work, Discussions with the supervisor, Journal club and Co-Curricular activities over the duration of four semesters. The credits 3 are distributed semester-wise as shown in Table 14. Courses generally progress in sequence, building competencies and their positioning indicates certain academic maturity on the part of the learners. Learners are expected to follow the semester-wise schedule of courses given in the syllabus.

## **10. Academic work**

A regular record of attendance both in Theory, Practical, Seminar, Assignment, Journal club, Discussion with the supervisor, Research work presentation and Dissertation shall be maintained by the department / teaching staff of respective courses.

## **11. Course of study**

The course of study for M. Pharm shall include Semester Wise Theory & Practical as given in Table– II-III. The number of hours to be devoted to each theory, and practical course in

any semester shall not be less than that shown in Table –II-III

**Table II:- Course of study for Semester I**

Course code	Name of the course	No. of hours	Tutorial	Credit Points
MPA101T	Modern Pharmaceutical Analytical Techniques	4	----	4
MPA102T	Advanced Pharmaceutical Analysis	4	----	4
MPA 103T	Pharmaceutical Validation	4	----	4
MPA 104T	Food Analysis	4	----	4
MPA 105P	Pharmaceutical Analysis Practical I	12	----	6
-	Seminar/Assignment	7	----	4
<b>Total</b>		<b>35</b>		<b>26</b>

**Table III:- Course of study for Semester II**

Course code	Name of the course	No. of hours	Tutorial	Credit Points
MPA201T	Advanced Instrumental Analysis	4	-	4
MPA 202T	Modern Bio-Analytical Techniques	4	-	4
MPA 203T	Quality Control and Quality Assurance	4	-	4
MPA 204T	Herbal and Cosmetic Analysis	4	-	4
MPA 205P	Pharmaceutical Analysis Practical II	12	-	6
	Seminar/Assignment	7	-	4
<b>Total</b>		<b>35</b>		<b>26</b>

**Table-IV: Semester wise credits distribution**

Semester	Credit Points
I	26
II	26
III	21
IV	20
Co-curricular Activities (Attending Conference, Scientific Presentations and Other Scholarly Activities)	Minimum=02 Maximum=07*
<b>Total credit points</b>	Minimum=95 Maximum=10*

\* Credit Points for Co-curricular Activities

## 12. Program Committee

1. The M. Pharm. programme shall have a Programme Committee constituted by the Head of the institution in consultation with all the Heads of the departments.

2. The composition of the Programme Committee shall be as follows: A teacher at the cadre of Professor shall be the Chairperson; One Teacher from each M. Pharm specialization and four student representatives (two from each academic year), nominated by the Head of the institution.
3. Duties of the Programme Committee:
  - i. Periodically reviewing the progress of the classes.
  - ii. Discussing the problems concerning curriculum, syllabus and the conduct of classes.
  - iii. Discussing with the course teachers on the nature and scope of assessment for the course and the same shall be announced to the students at the beginning of respective semesters.
  - iv. Communicating its recommendation to the Head of the institution on academic matters.
  - v. The Programme Committee shall meet at least twice in a semester preferably at the end of each sessional exam and before the end semester exam.

### 13. Examinations/Assessments

The scheme for internal assessment and end semester examinations is given in Table V- VIII

#### End semester examinations

The End Semester Examinations for each theory and practical course through semesters I to IV shall be conducted by the university except for the subjects with asterix symbol (\*) in table I and II for which examinations shall be conducted by the subject experts at college level and the marks/grades shall be submitted to the university.

#### Internal assessment: Continuous mode

The marks allocated for Continuous mode of Internal Assessment shall be awarded as per the scheme given below.

**Table V: Scheme for awarding internal assessment: Continuous mode**

<b>Theory</b>	
<b>Criteria</b>	<b>Maximum Marks</b>
Attendance (Refer Table – VI)	8
Student – Teacher interaction	2
<b>Total</b>	<b>10</b>
<b>Practical</b>	
Attendance (Refer Table – VI)	10
Based on Practical Records, Regular viva voce, etc.	10
<b>Total</b>	<b>20</b>

**Table VI: Guidelines for the allotment of marks for attendance**

<b>Percentage of Attendance</b>	<b>Theory</b>	<b>Practical</b>
95 – 100	8	10
90 – 94	6	7.5
85 – 89	4	5
80 – 84	2	2.5
Less than 80	0	0

### **Sessional Exams**

Two sessional exams shall be conducted for each theory / practical course as per the schedule fixed by the college(s). The scheme of question paper for theory and practical sessional examinations is given in the table. The average marks of two sessional exams shall be computed for internal assessment as per the requirements given in tables.

### **Promotion and award of grades**

A student shall be declared PASS and eligible for getting grade in a course of M.Pharm.programme if he/she secures at least 50% marks in that particular course including internal assessment.

### **Carry forward of marks**

In case a student fails to secure the minimum 50% in any Theory or Practical course as specified in 12, then he/she shall reappear for the end semester examination of that course. However his/her marks of the Internal Assessment shall be carried over and he/she shall be entitled for grade obtained by him/her on passing.

### **Improvement of internal assessment**

A student shall have the opportunity to improve his/her performance only once in the sessional exam component of the internal assessment. The re-conduct of the sessional exam shall be completed before the commencement of next end semester theory examinations.

### **Reexamination of end semester examinations**

Reexamination of end semester examination shall be conducted as per the schedule given in table VI. The exact dates of examinations shall be notified from time to time.

**Table VII: Tentative schedule of end semester examinations**

Semester	For Regular Candidates	For Failed Candidates
I and III	November / December	May / June
II and IV	May / June	November / December

**Allowed to keep terms (ATKT):**

No student shall be admitted to any examination unless he/she fulfills the norms given in 6. ATKT rules are applicable as follows:

A student shall be eligible to carry forward all the courses of I and II semesters till the III semester examinations. However, he/she shall not be eligible to attend the courses of IV semester until all the courses of I, II and III semesters are successfully completed.

A student shall be eligible to get his/her CGPA upon successful completion of the courses of I to IV semesters within the stipulated time period as per the norms.

Note: Grade AB should be considered as failed and treated as one head for deciding ATKT. Such rules are also applicable for those students who fail to register for examination(s) of any course in any semester.

**Grading of performances**

Letter grades and grade points allocations:

Based on the performances, each student shall be awarded a final letter grade at the end of the semester for each course. The letter grades and their corresponding grade points are given in Table – VIII.

**Table VIII: Letter grades and grade points equivalent to Percentage of marks and performances**

Percentage of Marks Obtained	Letter Grade	Grade Point	Performance
90.00 – 100	O	10	Outstanding
80.00 – 89.99	A	9	Excellent
70.00 – 79.99	B	8	Good
60.00 – 69.99	C	7	Fair
50.00 – 59.99	D	6	Average

Less than 50	F	0	Fail
Absent	AB	0	Fail

A learner who remains absent for any end semester examination shall be assigned a letter grade of AB and a corresponding grade point of zero. He/she should reappear for the said evaluation/examination in due course.

### **The Semester grade point average (SGPA)**

The performance of a student in a semester is indicated by a number called ‘Semester Grade Point Average’ (SGPA). The SGPA is the weighted average of the grade points obtained in all the courses by the student during the semester. For example, if a student takes five courses (Theory/Practical) in a semester with credits C<sub>1</sub>, C<sub>2</sub>, C<sub>3</sub> and C<sub>4</sub> and the student’s grade points in these courses

are G<sub>1</sub>, G<sub>2</sub>, G<sub>3</sub> and G<sub>4</sub>, respectively, and then students’ SGPA is equal to:

$$\text{SGPA} = \frac{C_1G_1 + C_2G_2 + C_3G_3 + C_4G_4}{C_1 + C_2 + C_3 + C_4}$$

The SGPA is calculated to two decimal points. It should be noted that, the SGPA for any semester shall take into consideration the F and ABS grade awarded in that semester. For example if a learner has a F or ABS grade in course 4, the SGPA shall then be computed as:

$$\text{SGPA} = \frac{C_1G_1 + C_2G_2 + C_3G_3 + C_4 * \text{ZERO}}{C_1 + C_2 + C_3 + C_4}$$

### **Cumulative Grade Point Average (CGPA)**

The CGPA is calculated with the SGPA of all the IV semesters to two decimal points and is indicated in final grade report card/final transcript showing the grades of all IV semesters and their courses. The CGPA shall reflect the failed status in case of F grade(s), till the course(s) is/are passed. When the course(s) is/are passed by obtaining a pass grade on subsequent examination(s) the CGPA shall only reflect the new grade and not the fail grades earned earlier. The CGPA is calculated as:

$$\text{CGPA} = \frac{C_1S_1 + C_2S_2 + C_3S_3 + C_4S_4}{C_1 + C_2 + C_3 + C_4}$$



where C1, C2, C3,.... is the total number of credits for semester I,II,III,.... and S1,S2, S3,....is the SGPA of semester I,II,III,.... .

### **Declaration of class**

The class shall be awarded on the basis of CGPA as follows: First Class with Distinction = CGPA of. 7.50 and above

First Class = CGPA of 6.00 to 7.49

Second Class = CGPA of 5.00 to 5.99

### **Project work**

All the students shall undertake a project under the supervision of a teacher in Semester III to IV and submit a report. 4 copies of the project report shall be submitted (typed & bound copy not less than 75 pages).

The internal and external examiner appointed by the University shall evaluate the project at the time of the Practical examinations of other semester(s). The projects shall be evaluated as per the criteria given below.

### **Evaluation of Dissertation Book:**

Objective(s) of the work done	50 Marks
Methodology adopted	150 Marks
Results and Discussions	250 Marks
Conclusions and Outcomes	50 Marks
Total	<u>500 Marks</u>

### **Evaluation of Presentation:**

Presentation of work	100 Marks
Communication skills	50 Marks
Question and answer skills	100 Marks
Total	250 Marks

### **Award of Ranks**

Ranks and Medals shall be awarded on the basis of final CGPA. However, candidates who fail in one or more courses during the M.Pharm program shall not be eligible for award of ranks. Moreover, the candidates should have completed the M. Pharm program in minimum prescribed number of years, (two years) for the award of Ranks.

**Award of degree**

Candidates who fulfill the requirements mentioned above shall be eligible for award of degree during the ensuing convocation.

**Duration for completion of the program of study**

The duration for the completion of the program shall be fixed as double the actual duration of the program and the students have to pass within the said period, otherwise they have to get freshRegistration.

**Revaluation I Re-totaling of answer papers**

There is no provision for revaluation of the answer papers in any examination. However, the candidates can apply for retotaling by paying prescribed fee.

**Re-admission after break of study**

Candidate who seeks re-admission to the program after break of study has to get the approval from the university by paying a condonationfee.

# **SYLLABUS**

# **M.PHARM. SEMESTER I**

<b>Course Code MPA 101T</b>	<b>Title of the course: Modern Pharmaceutical Analytical Techniques (MAT)</b>
<b>Course Code: MPA 102T</b>	<b>Title of the Course: Advanced Pharmaceutical Analysis (Theory)</b>
<b>Course Code: MPA 103T</b>	<b>Title of the Course: Pharmaceutical Validation (Theory)</b>
<b>Course Code: MPA 104T</b>	<b>Title of the Course: Food Analysis (Theory)</b>
<b>Course Code: MPA 105P</b>	<b>Title of the Course: Pharmaceutical Analysis Practical – I</b>

**Name of the Academic Program: M. Pharm. (Pharmaceutical Analysis)**

**Course Code: MPA101T**

**Title of the Course: Modern Pharmaceutical Analytical Techniques (Theory)**

**L-T 4 Credits: 4**

(L=Lecture hours, T=Tutorial hours)

### **COURSE OUTCOMES (COs)**

**After completing this Course, the students should be able to:**

**CO1:** Recognize the principle, instrumentation and applications of different chromatographic techniques (**Cognitive level: Understand**)

**CO2:** Investigate the pharmaceutical substance by Nuclear Magnetic spectroscopy techniques. (**Cognitive level: Apply**)

**CO3:** Investigate the pharmaceutical substance by Mass spectroscopy Techniques. (**Cognitive level: Apply**)

**CO4:** The analysis of various drugs in single and combination dosage forms (**Cognitive level: Create**)

**CO5:** Recognize the principle, instrumentation and applications of electrophoresis and X ray crystallography. (**Cognitive level: Apply**)

### **Mapping of Course Outcomes (COs) with Program Outcomes (POs) and Program Specific Outcomes (PSOs)**

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3	PSO4
<b>CO1</b>	1	2	2	3							1		2		
<b>CO2</b>	1	2	2	3							1		2		
<b>CO3</b>	1	2	2	3							1		2		
<b>CO4</b>	1	2	2	3							1		2		
<b>CO5</b>	1	2	2	3							1		2		

Each Course Outcome (CO) may be mapped with one or more Program Outcomes (POs). Write '3' in the box for 'High-level' mapping, 2 for 'Medium-level' mapping, 1 for 'Low'-level' mapping.

**Syllabus:**

**THEORY**

**60 Hrs**

1. **UV-Visible spectroscopy:** Introduction, Theory, Laws, 1 Instrumentation associated with UV-Visible spectroscopy, Choice of solvents and solvent effect and Applications of UV-Visible spectroscopy, Difference/ Derivative spectroscopy.

**IR spectroscopy:** Theory, Modes of Molecular vibrations, Sample handling, Instrumentation of Dispersive and Fourier - Transform IR Spectrometer, Factors affecting vibrational frequencies and Applications of IR spectroscopy, Data Interpretation.

**Spectrofluorimetry:** Theory of Fluorescence, Factors affecting fluorescence (Characteristics of drugs that can be analysed by fluorimetry), Quenchers, Instrumentation and Applications of fluorescence spectrophotometer.

**Flame emission spectroscopy and Atomic absorption spectroscopy:** Principle, Instrumentation, Interferences and Applications. (10Hrs)

2. **NMR spectroscopy:** Quantum numbers and their role in NMR, Principle, Instrumentation, Solvent requirement in NMR, Relaxation process, NMR signals in various compounds, Chemical shift, Factors influencing chemical shift, Spin-Spin coupling, Coupling constant, Nuclear magnetic double resonance, Brief outline of principles of FT-NMR and <sup>13</sup>C NMR. Applications of NMR spectroscopy. (10 Hrs)
3. **Mass Spectroscopy:** Principle, Theory, Instrumentation of Mass Spectroscopy, Different types of ionization like electron impact, chemical, field, FAB and MALDI, APCI, ESI, APPI Analyzers of Quadrupole and Time of Flight, Mass fragmentation and its rules, Meta stable ions, Isotopic peaks and Applications of Mass spectroscopy. (10 Hrs)
4. **Chromatography:** Principle, apparatus, instrumentation, chromatographic parameters, factors affecting resolution, isolation of drug from excipients, data interpretation and applications of the following: (10 Hrs)
  - a) Thin Layer chromatography
  - b) High Performance Thin Layer Chromatography
  - c) Ion exchange chromatography
  - d) Column chromatography
  - e) Gas chromatography
  - f) High Performance Liquid chromatography
  - g) Ultra High Performance Liquid chromatography
  - h) Affinity chromatography
  - i) Gel Chromatography
5. **Electrophoresis:** Principle, Instrumentation, Working conditions, factors affecting separation and applications of the following: (10 Hrs)
  - a) Paper electrophoresis
  - b) Gel electrophoresis
  - c) Capillary electrophoresis
  - d) Zone electrophoresis
  - e) Moving boundary electrophoresis
  - f) Isoelectric focusingX ray Crystallography: Production of X rays, Different X ray methods, Bragg 's law, Rotating crystal technique, X ray powder technique, Types of crystals and applications of X-ray diffraction.
6. **a. Potentiometry:** Principle, working, Ion selective Electrodes and Application of potentiometry.  
**b. Thermal Techniques:** Principle, thermal transitions and Instrumentation (Heat flux and power-compensation and designs), Modulated DSC, Hyper DSC, experimental parameters (sample preparation, experimental conditions, calibration, heating and cooling rates, resolution, source of errors) and their influence, advantage and disadvantages, pharmaceutical applications. Differential Thermal Analysis (DTA):

Principle, instrumentation and advantage and disadvantages, pharmaceutical applications, derivative differential thermal analysis (DDTA). TGA: Principle, instrumentation, factors affecting results, advantage and disadvantages, pharmaceutical applications. (10 Hrs)

#### REFERENCES:

1. Spectrometric Identification of Organic compounds - Robert M Silverstein, Sixth edition, John Wiley & Sons, 2004.
2. Principles of Instrumental Analysis - Douglas A Skoog, F. James Holler, Timothy A. Nieman, 5th edition, Eastern press, Bangalore, 1998.
3. Instrumental methods of analysis – Willards, 7th edition, CBS publishers.
4. Practical Pharmaceutical Chemistry – Beckett and Stenlake, Vol II, 4th edition, CBS Publishers, New Delhi, 1997.
5. Organic Spectroscopy - William Kemp, 3rd edition, ELBS, 1991.
6. Quantitative Analysis of Drugs in Pharmaceutical formulation - P D Sethi, 3rd Edition, CBS Publishers, New Delhi, 1997.
7. Pharmaceutical Analysis - Modern Methods – Part B - J W Munson, Vol 11, Marcel. Dekker Series
8. Spectroscopy of Organic Compounds, 2nd edn., P.S/Kalsi, Wiley eastern Ltd., Delhi.
9. Textbook of Pharmaceutical Analysis, KA.Connors, 3rd Edition, John Wiley & Sons, 1982.

#### Teaching-Learning Strategies in Brief

The teaching learning strategies, followed are board and chalk teaching, Learning through discussion among the peer group, classroom interaction, discussion of research papers of Journals related to topics, power point presentation, Q & A session and reflective learning.

#### Assessment methods and weightages in brief

There are two components of assessment: Internal assessment (25 marks) and End semester examination (75 marks). Internal assessment consists of continuous mode (10 marks) and sessional examinations (15 marks). There are two Sessional exams (each conducted for 30 marks and computed for 15 marks) and one improvement exam. The average marks of two best sessional exams are computed out of 15 marks. Continuous mode evaluation is of 10 marks comprising of attendance- 8 marks (calculated as: Percentage of Attendance: Allotment of marks- Less than 80: 0 marks; 80-84: 2 marks; 85-89: 4 marks; 90-94: 6 marks and 95-100: 8 marks) and Student- Teacher interaction: 2 marks.

Total Marks are 100 for the subject (Internal Assessment: 25 marks and End Semester Examination: 75 Marks).

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**Name of the Academic Program: M. Pharm (Pharmaceutical Analysis)**

**Course Code: MPA 102T**

**Title of the Course: Advanced Pharmaceutical Analysis (Theory)**

**L- P: 4-0**

**Credits: 4**

(L=Lecture hours, P=Practical hours)

### **COURSE OUTCOMES (COs)**

After completing this Course, the students should be able to:

**CO-1:** Develop analytical skills required for analytical method development (Cognitive level: Apply)

**CO-2:** Discuss principles of various reagents used in functional group analysis that renders necessary support in research methodology and demonstrates its application in practical related problems. (Cognitive level: Understand)

**CO-3:** Analyse impurities in drugs, residual solvents and stability studies of drugs and biological products. (Cognitive level: Analyse)

**CO-4:** Utilize principles and procedures for biological testing of various vaccines. (Cognitive level: Apply)

**CO-5:** Know stability testing of phytopharmaceuticals and their protocol preparation. (Cognitive level: Understand)

### **Mapping of Course Outcomes (COs) with Program Outcomes (POs) and Program Specific Outcomes (PSOs)**

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3	PSO4
<b>CO1</b>	3	3	3	3		2				2	2	3	3	2	
<b>CO2</b>	3	2	3	3		1				3	1	3	3	2	
<b>CO3</b>	3	2	2	3			1				1	3	3	3	2
<b>CO4</b>	3		3				2	1	2	3		3		3	3
<b>CO5</b>	3	2	3	3	1	1	2	1	2	3		3	2	3	3

Each Course Outcome (CO) may be mapped with one or more Program Outcomes (POs). Write '3' in the box for 'High-level' mapping, 2 for 'Medium-level' mapping, 1 for 'Low'-level' mapping.

### **Detailed Syllabus**

**60 Hrs**

#### **1. Impurity and stability studies:**

**10 Hrs**

Definition, classification of impurities in drug Substance or Active Pharmaceutical Ingredients and quantification of impurities as per ICH guidelines

Impurities in new drug products:



Rationale for the reporting and control of degradation products, reporting degradation products content of batches, listing of degradation products in specifications, qualification of degradation products

Impurities in residual solvents:

General principles, classification of residual solvents, Analytical procedures, limits of residual solvents, reporting levels of residual solvents

**2. Elemental impurities: 10 Hrs**

Element classification, control of elemental impurities, Potential Sources of elemental Impurities, Identification of Potential Elemental Impurities, analytical procedures, instrumentation & C, H, N and S analysis

Stability testing protocols:

Selection of batches, container orientation, test parameters, sampling frequency, specification, storage conditions, recording of results, concept of stability, commitment etc. Important mechanistic and stability related information provided by results of study of factors like temperature, pH, buffering species ionic strength and dielectric constant etc. on the reaction rates. With practical considerations.

**3. Impurity profiling and degradant characterization: 10 Hrs**

Method development, Stability studies and concepts of validation accelerated stability testing & shelf life calculation, WHO and ICH stability testing guidelines, Stability zones, steps in development, practical considerations. Basics of impurity profiling and degradant characterization with special emphasis. Photostability testing guidelines, ICH stability guidelines for biological products

**4. Stability testing of phytopharmaceuticals: 10 Hrs**

Regulatory requirements, protocols, HPTLC/HPLC finger printing, interactions and complexity.

**5. Biological tests and assays of the following: 10 Hrs**

- a. Adsorbed Tetanus vaccine
- b. Adsorbed Diphtheria vaccine
- c. Human anti haemophilic vaccine
- d. Rabies vaccine
- e. Tetanus Anti toxin
- f. Tetanus Anti serum
- g. Oxytocin
- h. Heparin sodium IP
- i. Antivenom. PCR, PCR studies for gene regulation, instrumentation (Principle and Procedures)

**6. Immunoassays (IA) 10 Hrs**

Basic principles, Production of antibodies, Separation of bound and unbound drug,

Radioimmunoassay, Optical IA, Enzyme IA, Fluoro IA, Luminiscence IA, Quantification and applications of IA.

### **Reference Books:**

1. Vogel's textbook of quantitative chemical analysis - Jeffery J Bassett, J. Mendham, R. C. Denney, 5th edition, ELBS, 1991.
2. Practical Pharmaceutical Chemistry - Beckett and Stenlake, Vol II, 4th Edition, CBS publishers, New Delhi, 1997.
3. Textbook of Pharmaceutical Analysis - K A Connors, 3rd Edition, John Wiley & Sons, 1982.
4. Pharmaceutical Analysis - Higuchi, Brochmman and Hassen, 2nd Edition, Wiley – Inter science Publication, 1961.
5. Quantitative Analysis of Drugs in Pharmaceutical formulation – P D Sethi, 3rd Edition, CBS Publishers New Delhi, 1997.
6. Pharmaceutical Analysis- Modern methods - J W Munson – Part B, Volume 11, Marcel Dekker Series.
7. The Quantitative analysis of Drugs - D C Carratt, 3rd edition, CBS Publishers, New Delhi, 1964.
8. Indian Pharmacopoeia Vol I , II & III 2007, 2010, 2014.
9. Methods of sampling and microbiological examination of water, first revision, BIS
10. Practical HPLC method development – Snyder, Kirkland, Glajch, 2nd edition, John Wiley & Sons.
11. Analytical Profiles of drug substances – Klaus Florey, Volume 1 – 20, Elsevier, 2005
12. Analytical Profiles of drug substances and Excipients – Harry G Brittan, Volume 21 – 30, Elsevier, 2005.
13. The analysis of drugs in biological fluids - Joseph Chamberlain, 2nd edition, CRC press, London.
14. ICH Guidelines for impurity profiles and stability studies.

### **Teaching-Learning Strategies in Brief**

The teaching learning strategies, followed are board and chalk teaching, Learning through discussion among the peer group, classroom interaction, discussion of research papers of Journals related to topics, power point presentation, Q & A session and reflective learning.

### **Assessment methods and weightages in brief**

There are two components of assessment: Internal assessment (25 marks) and End semester examination (75 marks). Internal assessment consists of continuous mode (10 marks) and sessional examinations (15 marks). There are two Sessional exams (each conducted for 30 marks and computed for 15 marks) and one improvement exam. The average marks of two best sessional exams are computed out of 15 marks. Continuous mode evaluation is of 10 marks comprising of attendance- 8 marks (calculated as: Percentage of Attendance: Allotment of marks- Less than 80: 0 marks; 80-84: 2 marks; 85-89: 4 marks; 90-94: 6 marks and 95-100: 8 marks) and Student- Teacher interaction: 2 marks.

Total Marks are 100 for the subject (Internal Assessment: 25 marks and End Semester Examination: 75 Marks).

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**Name of the Academic Program: M. Pharm. (Pharmaceutical Analysis)**

**Course Code: MPA-103T**

**Title of the Course: Pharmaceutical Validation (Theory)**

**L-T-P: 4-0-12**

**Credits: 4**

(L=Lecture hours, T=Tutorial hours, P=Practical hours)

### **COURSE OUTCOMES (COs)**

After the course has been completed students should be able to:

**CO-1:** Explain different aspects of validation (Cognitive level: Apply)

**CO-2:** Understand validation of instruments and equipments (Cognitive level: Understand)

**CO-3:** Understand validation of manufacturing processes (Cognitive level: Understand)

**CO-4:** Validate the manufacturing facilities (Cognitive level: Analyse)

### **Mapping of Course Outcomes (COs) with Program Outcomes (POs) and Program Specific Outcomes (PSOs)**

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO 11	PSO1	PSO2	PSO3	PSO4
<b>CO1</b>	3	3	2	3	3		2	1	2		3	3	1		3
<b>CO2</b>	3	2	2	2		2	2	1	2		2	3	1	2	2
<b>CO3</b>	3	2	2	2		3	2		2	2	2	3	1	2	2
<b>CO4</b>	3	1	1	1	3		2		3	2	2	3	1	2	2

Each Course Outcome (CO) may be mapped with one or more Program Outcomes (POs). Write '3' in the box for 'High-level' mapping, 2 for 'Medium-level' mapping, 1 for 'Low'-level' mapping.

### **Detailed Syllabus**

**60 Hrs**

- 1. Introduction:** Definition of Qualification and Validation, Advantage of Validation, Streamlining of Qualification & Validation process and Validation Master Plan. Qualification: User Requirement Specification, Design Qualification, Factory Acceptance Test (FAT)/ Site Acceptance Test (SAT), Installation Qualification, Operational Qualification, Performance Qualification, Re- Qualification (Maintaining status- Calibration Preventive Maintenance, Change management), Qualification of Manufacturing Equipments, Qualification of Analytical Instruments and Laboratory equipments. **12 Hrs**
- 2. Qualification of analytical instruments:** Electronic balance, pH meter, UV-Visible spectrophotometer, FTIR, GC, HPLC, HPTLC. Qualification of Glassware: Volumetric flask, pipette, Measuring cylinder, beakers and burette. **12 Hrs**
- 3. Validation of Utility systems:** Pharmaceutical Water System & pure steam, HVAC system, Compressed air and nitrogen. Cleaning Validation: Cleaning Validation -

Cleaning Method development, Validation and validation of analytical method used in cleaning. Cleaning of Equipment, Cleaning of Facilities. Cleaning in place (CIP).

**12 Hrs**

**4 Analytical method validation:** General principles, Validation of analytical method as per ICH guidelines and USP. Computerized system validation: Electronic records and digital significance-21 CFR part 11 and GAMP 5. **12 Hrs**

**5 General Principles of Intellectual Property:** Concepts of Intellectual Property (IP), Intellectual Property Protection (IPP), Intellectual Property Rights (IPR); Economic importance, mechanism for protection of Intellectual Property –patents, Copyright, Trademark; Factors affecting choice of IP protection; Penalties for violation; Role of IP in pharmaceutical industry; Global ramification and financial implications. Filing a patent applications; patent application forms and guidelines. Types patent applications-provisional and non-provisional, PCT and convention patent applications; International patenting requirement procedures and costs; Rights and responsibilities of a patentee; Practical aspects regarding maintaining of a Patent file; Patent infringement meaning and scope. Significance of transfer technology (TOT), IP and ethics-positive and negative aspects of IPP; Societal responsibility, avoiding unethical practices. **12 Hrs**

#### **Reference Books:**

1. B. T. Loftus & R. A. Nash, "Pharmaceutical Process Validation", Drugs and Pharm Sci. Series, Vol. 129, 3rd Ed., Marcel Dekker Inc., N.Y.
2. The Theory & Practice of Industrial Pharmacy, 3rd edition, Leon Lachman, Herbert A. Lieberman, Joseph. L. Karig, Varghese Publishing House, Bombay.
3. Validation Master plan by Terveeks or Deeks, Davis Harwood International publishing.
4. Validation of Aseptic Pharmaceutical Processes, 2nd Edition, by Carleton & Agalloco, (Marcel Dekker).
5. Michael Levin, Pharmaceutical Process Scale-Up, Drugs and Pharm. Sci. Series, Vol. 157, 2nd Ed., Marcel Dekker Inc., N.Y.
6. Validation Standard Operating Procedures: A Step by Step Guide for Achieving Compliance in the Pharmaceutical, Medical Device, and Biotech Industries, Syed Imtiaz Haider.
7. Pharmaceutical Equipment Validation: The Ultimate Qualification Handbook, Phillip A. Cloud, Interpharm Press.
8. Validation of Pharmaceutical Processes: Sterile Products, Frederick J. Carlton (Ed.) and James Agalloco (Ed.), Marcel Dekker, 2nd Ed.
9. Analytical Method validation and Instrument Performance Verification by Churg Chan, Heiman Lam, Y.C. Lee, Yue. Zhang, Wiley Inter Science.

### **Teaching-Learning Strategies in Brief**

The teaching learning strategies, followed are board and chalk teaching, Learning through discussion among the peer group, classroom interaction, discussion of research papers of Journals related to topics, power point presentation, Q & A session and reflective learning.

### **Assessment methods and weightages in brief**

There are two components of assessment: Internal assessment (25 marks) and End semester examination (75 marks). Internal assessment consists of continuous mode (10 marks) and sessional examinations (15 marks). There are two Sessional exams (each conducted for 30 marks and computed for 15 marks) and one improvement exam. The average marks of two best sessional exams are computed out of 15 marks. Continuous mode evaluation is of 10 marks comprising of attendance- 8 marks (calculated as: Percentage of Attendance: Allotment of marks- Less than 80: 0 marks; 80-84: 2 marks; 85-89: 4 marks; 90-94: 6 marks and 95-100: 8 marks) and Student- Teacher interaction: 2 marks.

Total Marks are 100 for the subject (Internal Assessment: 25 marks and End Semester Examination: 75 Marks).

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**Name of the Academic Program: M. Pharm. (Pharmaceutical Analysis)**

**Course Code: MPA-104T**

**Title of the Course: Food Analysis (Theory)**

**L-T-P: 4-0-0**

**Credits: 4**

(L=Lecture hours, T=Tutorial hours, P=Practical hours)

### **COURSE OUTCOMES (COs)**

Upon course completion, students should be able to:

**CO-1:** Explain the criteria for the analysis of food constituents and additives (**Cognitive level: Understand**)

**CO-2:** Understand Pesticides, its effect on foods and agriculture. (**Cognitive level: Understand**)

**CO-3:** Learn about various food regulations and legislations. (**Cognitive level: Understand**)

**CO-4:** Have knowledge regarding food carbohydrates, lipids, vitamins and proteins (**Cognitive level: Understand**)

**CO-5:** Use the concept in analysis of various food products (**Cognitive level: Apply**)

### **Mapping of Course Outcomes (COs) with Program Outcomes (POs) and Program Specific Outcomes (PSOs)**

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3	PSO4
<b>CO1</b>	3	1	3	1	1	1	2	1	1	2	3		3		
<b>CO2</b>	3	1	3	1	1	1	2	1	1	2	2		3		
<b>CO3</b>	3	1	3	2	1	1	2	1	1	2	2		3		
<b>CO4</b>	3	1	3	2	1	1	2	1	1	2	2		3		
<b>CO5</b>	3	1	3	3	1	1	2	1	1	2	2		3		

Each Course Outcome (CO) may be mapped with one or more Program Outcomes (POs). Write '3' in the box for 'High-level' mapping, 2 for 'Medium-level' mapping, 1 for 'Low'-level' mapping.

### **Detailed Syllabus**

**60 Hrs**

#### **Unit-I**

**12 Hours**

**Carbohydrates:** classification and properties of food carbohydrates, General methods of analysis of food carbohydrates, Changes in food carbohydrates during processing, Digestion, absorption and metabolism of carbohydrates, Dietary fibre, Crude fibre and application of food carbohydrates.

**Proteins:** Chemistry and classification of amino acids and proteins, Physico-Chemical properties of protein and their structure, general methods of analysis of proteins and amino acids, Digestion, absorption and metabolism of proteins.

**Unit-II****12 Hours**

**Lipids:** Classification, general methods of analysis, refining of fats and oils; hydrogenation of vegetable oils, Determination of adulteration in fats and oils, Various methods used for measurement of spoilage of fats and fatty foods.

**Vitamins:** classification of vitamins, methods of analysis of vitamins, Principles of microbial assay of vitamins of B-series.

**Unit-III****12 Hours**

**Food additives:** Introduction, analysis of Preservatives, antioxidants, artificial sweeteners, flavors, flavor enhancers, stabilizers, thickening and jelling agents.

Pigments and synthetic dyes: Natural pigments, their occurrence and characteristic properties, permitted synthetic dyes, Non-permitted synthetic dyes used by industries, Method of detection of natural, permitted and non-permitted dyes.

**Unit-IV****12 Hours**

General Analytical methods for milk, milk constituents and milk products like ice cream, milk powder, butter, margarine, cheese including adulterants and contaminants of milk.

Analysis of fermentation products like wine, spirits, beer and vinegar.

**Unit-V****12 Hours**

**Pesticide analysis:** Effects of pest and insects on various food, use of pesticides in agriculture, pesticide cycle, organophosphorus and organochlorine pesticides analysis, determination of pesticide residues in grain, fruits, vegetables, milk and milk products.

Legislation regulations of food products with special emphasis on BIS, Agmark, FDA and US-FD

**Reference Books:**

1. The chemical analysis of foods – David Pearson, Seventh edition, Churchill Livingstone, Edinburgh London, 1976
2. Introduction to the Chemical analysis of foods – S. Nielsen, Jones & Bartlett publishers, Boston London, 1994.
3. Official methods of analysis of AOAC International, sixth edition, Volume I & II, 1997.
4. Analysis of Food constituents – Multon, Wiley VCH.
5. Dr. William Horwitz, Official methods of analysis of AOAC International, 18th edition, 2005.

**Teaching-Learning Strategies in Brief**

The teaching learning strategies, followed are board and chalk teaching, Learning through discussion among the peer group, classroom interaction, discussion of research papers of Journals related to topics, power point presentation, Q & A session and reflective learning.



### **Assessment methods and weightages in brief**

There are two components of assessment: Internal assessment (25 marks) and End semester examination (75 marks). Internal assessment consists of continuous mode (10 marks) and sessional examinations (15 marks). There are two Sessional exams (each conducted for 30 marks and computed for 15 marks) and one improvement exam. The average marks of two best sessional exams are computed out of 15 marks. Continuous mode evaluation is of 10 marks comprising of attendance- 8 marks (calculated as: Percentage of Attendance: Allotment of marks- Less than 80: 0 marks; 80-84: 2 marks; 85-89: 4 marks; 90-94: 6 marks and 95-100: 8 marks) and Student- Teacher interaction: 2 marks.

Total Marks are 100 for the subject (Internal Assessment: 25 marks and End Semester Examination: 75 Marks).

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**Name of the Academic Program: M. Pharm. (Pharmaceutical Analysis)**

**Course Code: MPA-105T**

**Title of the Course: Pharmaceutical Analytical Techniques Practical's -I**

**L-T-P: 0-0-12**

**Credits: 6**

(L=Lecture hours, T=Tutorial hours, P=Practical hours)

### **COURSE OUTCOMES (COs)**

Upon course completion, students should be able to:

**CO-1:** Understand the principle and theory involved in the various instrumental techniques  
(**Cognitive level: Understand and Apply**)

**CO-2:** Acquire the knowledge and practical skills required to analyse drugs (**Cognitive level: Apply**)

**CO-3:** Interpret the data obtained in various spectroscopic methods (**Cognitive level: Apply**)

**CO-4:** Apply instrumental and non-instrumental techniques in the analysis of different formulations (**Cognitive level: Understand**)

**CO-5:** Perform qualitative and quantitative analysis of pharmaceuticals using various analytical techniques (**Cognitive level: Understand**)

### **Mapping of Course Outcomes (COs) with Program Outcomes (POs) and Program Specific Outcomes (PSOs)**

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3	PSO4
<b>CO1</b>	3			2					3			3			
<b>CO2</b>	3										3	3			
<b>CO3</b>	3	2	3	2					1	2	3	3			
<b>CO4</b>	3		3							2	3	3			
<b>CO5</b>	3	3	2	3					2	2	3	3			

Each Course Outcome (CO) may be mapped with one or more Program Outcomes (POs). Write '3' in the box for 'High-level' mapping, 2 for 'Medium-level' mapping, 1 for 'Low'-level' mapping.

### **Detailed Syllabus**

**60 Hrs**

1. Analysis of Pharmacopeial compounds and their formulations by UV Vis spectrophotometer
2. Simultaneous estimation of multi component containing formulations by UV spectrophotometry
3. Experiments based on HPLC
4. Experiments based on Gas Chromatography
5. Estimation of riboflavin/quinine sulphate by fluorimetry
6. Estimation of sodium/potassium by flame photometry
7. Assay of official compounds by different titrations

8. Assay of official compounds by instrumental techniques.
9. Quantitative determination of hydroxyl group.
10. Quantitative determination of amino group
11. Colorimetric determination of drugs by using different reagents
12. Impurity profiling of drugs
13. Calibration of glasswares
14. Calibration of pH meter
15. Calibration of UV-Visible spectrophotometer
16. Calibration of FTIR spectrophotometer
17. Calibration of GC instrument
18. Calibration of HPLC instrument
19. Cleaning validation of any one equipment
20. Determination of total reducing sugar
21. Determination of proteins
22. Determination of saponification value, Iodine value, Peroxide value in food products
23. Determination of fat content and rancidity in food products
24. Analysis of natural and synthetic colors in food
25. Determination of preservatives in food
26. Determination of pesticide residue in food products
27. Analysis of vitamin content in food products
28. Determination of density and specific gravity of foods
29. Determination of food additives

### **Reference Books**

1. The chemical analysis of foods – David Pearson, Seventh edition, Churchill Livingstone, Edinburgh London, 1976
2. Introduction to the Chemical analysis of foods – S. Nielsen, Jones & Bartlett publishers, Boston London, 1994.
3. Official methods of analysis of AOAC International, sixth edition, Volume I & II, 1997.
4. B. T. Loftus & R. A. Nash, "Pharmaceutical Process Validation", Drugs and Pharm Sci. Series, Vol. 129, 3rd Ed., Marcel Dekker Inc., N.Y.
5. Validation Master plan by Terveeks or Deeks, Davis Harwood International publishing.
6. Principles of Instrumental Analysis - Douglas A Skoog, F. James Holler, Timothy A. Nieman, 5th edition, Eastern press, Bangalore, 1998.
7. Instrumental methods of analysis – Willards, 7th edition, CBS publishers.
8. Practical Pharmaceutical Chemistry – Beckett and Stenlake, Vol II, 4<sup>th</sup> edition, CBS Publishers, New Delhi, 1997.
9. Organic Spectroscopy - William Kemp, 3rd edition, ELBS, 1991.
10. Quantitative Analysis of Drugs in Pharmaceutical formulation - P D Sethi, 3rd Edition, CBS Publishers, New Delhi, 1997.

## **Teaching-Learning Strategies in brief**

The teaching learning strategies, followed are chalk-board teaching, learning through discussion among the peer group, classroom interaction, quiz, presentations, Q & A session and reflective learning.

## **Assessment methods and weightages in brief**

### **Practical**

*There are two components of assessment:* Internal assessment (50 marks) and End semester examination (100 marks). Internal assessment consists of continuous mode (20 marks) and sessional examinations (30 marks). Continuous mode evaluation is of 10 marks comprising of Attendance- 10 marks (calculated as: Percentage of Attendance: Allotment of marks- Less than 80: 0 marks; 80-84: 2.5 mark; 85-89:5 mark; 90-94: 7.5 marks and 95-100: 10 marks) and based on practical records, regular viva voce, etc. -10 marks. There are two Sessional exams (each conducted for 40 marks and computed for 30 marks) and one improvement exam (40 marks and computed for 30 marks). The average marks of two best sessional exams are computed out of 30 marks.

Total Marks are 150 for the subject (Internal Assessment: 50 marks and End Semester Examination: 100 Marks)

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# **M.PHARM. SEMESTER II**

<b>Course Code MPA 201T</b>	<b>Title of the course: Advanced Spectral Analysis</b>
<b>Course Code: MPA 202T</b>	<b>Title of the Course: Modern Bioanalytical Techniques</b>
<b>Course Code: MPA 203T</b>	<b>Title of the Course: Quality Control and Quality Assurance</b>
<b>Course Code: MPA 204T</b>	<b>Title of the Course: Herbal and Cosmetic Analysis</b>
<b>Course Code: MPA 205P</b>	<b>Title of the Course: Pharmaceutical Analysis Practical-II</b>

**Name of the Academic Program: M. Pharm. (Pharmaceutical Analysis)**

**Course Code: MPA-201T**

**Title of the Course: Advanced Spectral Analysis (Theory)**

**L-T-P: 4-0-12**

**Credits: 4**

(L=Lecture hours, T=Tutorial hours, P=Practical hours)

### **COURSE OUTCOMES (COs)**

After completing this Course, the students should be able to understand:

**CO-1:** Identification of organic compounds using instrumental techniques (**Cognitive level: Analyse**)

**CO-2:** About the theory and practical of the instruments used in analysis (**Cognitive level: Understand**)

**CO-3:** Chromatographic analysis of various compounds and its validation techniques (**Cognitive level: Analyse**)

**CO-4:** The principal instrumentation and applications of hyphenated instruments like LC-MS/MS, GC-MS/MS (**Cognitive level: Apply**)

**CO-5:** Spectral interpretation of NMR and Mass spectra of various organic compounds (**Cognitive level: Analyse**)

### **Mapping of Course Outcomes (COs) with Program Outcomes (POs) and Program Specific Outcomes (PSOs)**

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO 11	PSO1	PSO2	PSO3	PSO4
<b>CO1</b>	3	3	3	3	1	1	1	1	1	1	2	3	3	2	2
<b>CO2</b>	1	2	2	3	1	1	1	1	2	2	2	3	3	3	3
<b>CO3</b>	3	2	2	3	1	1	1	1	1	2	2	3	3	3	2
<b>CO4</b>	1	2	3	3	2	1	1	1	1	2	3	1	1	1	2
<b>CO5</b>	2	2	3	3	1	1	2	1	1	1	3	3	3	1	1

Each Course Outcome (CO) may be mapped with one or more Program Outcomes (POs). Write '3' in the box for 'High-level' mapping, 2 for 'Medium-level' mapping, 1 for 'Low'-level' mapping.

### **Detailed Syllabus**

**60 Hrs**

#### **Unit-I**

**12 Hrs**

**HPLC:** Principle, instrumentation, pharmaceutical applications, peak shapes, capacity factor, selectivity, plate number, plate height, resolution, band broadening, pumps, injector, detectors, columns, column problems, gradient HPLC, HPLC solvents, trouble shooting, sample preparation, method development, New developments in HPLC-role and principles of ultra, nano liquid chromatography in pharmaceutical analysis. Immobilized polysaccharide CSP's: Advancement in enantiomeric separations, revised phase Chiral method development and

HILIC approaches. HPLC in Chiral analysis of pharmaceuticals. Preparative HPLC, practical aspects of preparative HPLC.

## **Unit-II**

**12 Hrs**

**Biochromatography:** Size exclusion chromatography, ion exchange chromatography, ion pair chromatography, affinity chromatography general principles, stationary phases and mobile phases. **Gas chromatography:** Principles, instrumentation, derivatization, head space sampling, columns for GC, detectors, quantification. **High performance Thin Layer chromatography:** Principles, instrumentation, pharmaceutical applications.

## **Unit-III**

**12 Hrs**

**Super critical fluid chromatography:** Principles, instrumentation, pharmaceutical applications. Capillary electrophoresis: Overview of CE in pharmaceutical analysis, basic configuration, CE characteristics, principles of CE, methods and modes of CE. General considerations and method development in CE, Crown ethers as buffer additives in capillary electrophoresis. CE-MS hyphenation.

## **Unit-IV**

**12 Hrs**

**Mass spectrometry:** Principle, theory, instrumentation of mass spectrometry, different types of ionization like electron impact, chemical, field, FAB and MALD, APCI, ESI, APPI mass fragmentation and its rules, meta stable ions, isotopic peaks and applications of mass spectrometry. LC-MS hyphenation and DART MS analysis. Mass analysers (Quadrupole, Time of flight, FT-ICR, ion trap and Orbitrap) instruments. MS/MS systems (Tandem: QqQ, TOF-TOF; Q-IT, Q-TOF, LTQ-FT, LTQ-Orbitrap).

## **Unit-V**

**12 Hrs**

**NMR spectroscopy:** Quantum numbers and their role in NMR, Principle, Instrumentation, Solvent requirement in NMR, Relaxation process, NMR signals in various compounds, Chemical shift, Factors influencing chemical shift, Spin-Spin coupling, Coupling constant, Nuclear magnetic double resonance, Brief outline of principles of FT-NMR with reference to <sup>13</sup>CNMR: Spin spin and spin lattice relaxation phenomenon. <sup>13</sup>C NMR, 1-D and 2-D NMR, NOESY and COSY techniques, Interpretation and Applications of NMR spectroscopy. LC-NMR hyphenations.

## **Reference Books:**

1. Spectrometric Identification of Organic compounds - Robert M Silverstein, 8<sup>th</sup> edition, John Wiley & Sons, 2014.
2. Principles of Instrumental Analysis - Douglas A Skoog, F. James Holler, Timothy A. Nieman, 7th edition, Eastern press, Bangalore, 2017.
3. Instrumental methods of analysis – Willards, 7th edition, CBS publishers.
4. Organic Spectroscopy - William Kemp, 3rd edition, ELBS, 1991.
5. Quantitative analysis of Pharmaceutical formulations by HPTLC - P D Sethi, CBS Publishers, New Delhi.

6. Quantitative Analysis of Drugs in Pharmaceutical formulation - P D Sethi, 3 rd Edition, CBS Publishers, New Delhi, 1997.
7. Pharmaceutical Analysis- Modern methods – Part B - J W Munson, Volume 11, Marcel Dekker Series.
8. Organic Spectroscopy by Donald L. Paviya, 5th Edition.

### **Teaching-Learning Strategies in brief**

The teaching learning strategies, followed are chalk-board teaching, learning through discussion among the peer group, classroom interaction, quiz, presentations, Q & A session and reflective learning.

### **Assessment methods and weightages in brief**

There are two components of assessment: Internal assessment (25 marks) and End semester examination (75 marks). Internal assessment consists of continuous mode (10 marks) and sessional examinations (15 marks). Continuous mode evaluation is of 10 marks comprising of Attendance -8 marks (calculated as: Percentage of Attendance: Allotment of marks- Less than 80: 0 marks; 80-84: 2 mark; 85-89: 4 marks; 90-94: 6 marks and 95-100: 8 marks), Academic activities

Student teacher interaction-2 marks. There are two Sessional exams (each conducted for 30 marks and computed for 15 marks) and one improvement exam (30 marks and computed for 15 marks). The average marks of two best sessional exams are computed out of 15 marks.

Total Marks are 100 for the subject (Internal Assessment: 25 marks and End Semester Examination: 75 Marks)

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**Name of the Academic Program: M. Pharm (Pharmaceutical Analysis)**

**Course Code: MPA 202T**

**Title of the Course: Modern Bioanalytical Techniques (Theory)**

**L-T-P: 4-0-0**

**Credits: 4**

(L=Lecture hours, T=Tutorial hours, P=Practical hours)

### **COURSE OUTCOMES (COs)**

After completing this Course, the students should be able to:

**CO-1:** Explain about the importance of analysis of drugs in biological matrices (**Cognitive level: Understand**)

**CO-2:** The subject provides enough knowledge to conduct bioequivalence studies (**Cognitive level: Evaluate**)

**CO-3:** Understanding the biopharmaceutical factors affecting the pharmacokinetic of drugs (**Cognitive level: Understand**)

**CO-4:** Employ the knowledge of separation techniques for isolation of drug molecules from biological samples by different techniques. (**Cognitive level: Create**)

**CO-5:** It upgrade the procedure of bioequivalence study for formulations as per the guidelines of regulatory bodies by utilizing the proper regulatory guidelines. (**Cognitive level: Evaluate**)

**CO- 6:** Understanding cell culture techniques and its use in metabolite identification of drugs (**Cognitive level: Apply**)

**CO-7:** Employ of hyphenated instrumental techniques for analysis of drug molecules in the biological samples (**Cognitive level: Create**)

**CO-8:** Employ the use of flow cytometry in cell culture (Cognitive level: **Evaluate**)

### **Mapping of Course Outcomes (COs) with Program Outcomes (POs) and Program Specific Outcomes (PSOs)**

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3	PSO4
<b>CO1</b>	3	2	3	3	2	2	2	1	2	2	2	3	3	2	2
<b>CO2</b>	3	2	3	3	2	2	2	1	2	2	2	3	3	2	2
<b>CO3</b>	3	2	3	3	2	2	2	1	2	2	2	3	3	2	2
<b>CO4</b>	3	2	3	3	2	2	2	1	2	2	2	3	3	2	2
<b>CO5</b>	3	2	3	3	2	2	2	1	2	2	2	3	3	2	2
<b>CO6</b>	3	2	3	3	2	2	2	1	2	2	2	3	3	2	2
<b>CO7</b>	3	2	3	3	2	2	2	1	2	2	2	3	3	2	2
<b>CO8</b>	3	2	3	3	2	2	2	1	2	2	2	3	3	2	2

Each Course Outcome (CO) may be mapped with one or more Program Outcomes (POs). Write '3' in the box for 'High-level' mapping, 2 for 'Medium-level' mapping, 1 for 'Low'-level' mapping.

**Detailed Syllabus****Theory****60 Hrs**

- 1. Extraction of Drugs and Metabolites from Biological Matrices:** General need, and procedure involved in the bio-analytical methods such as protein precipitation, liquid - liquid extraction and solid phase extraction and other novel sample preparation approach. Bio-analytical method validation: USFDA and EMEA guidelines. **12 Hrs**
- 2. Biopharmaceutical Consideration:** Introduction, biopharmaceutical factors affecting drug bioavailability, in vitro: Dissolution and drug release testing, alternative methods of dissolution testing, transport models, biopharmaceutics classification system. Solubility: Experimental methods. Permeability: In-vitro, in-situ and in-vivo methods. **12 Hrs**
- 3. Pharmacokinetics and Toxicokinetics:** Basic consideration, drug interaction (PK-PD interactions). The effect of protein-binding interactions, effect of tissue-binding interactions, Cytochrome P450-based drug interactions, drug interactions linked to transporters. Microsomal assays. Toxicokinetics- Toxicokinetic evaluation in preclinical studies, importance and applications of toxicokinetic studies. LC-MS in bioactivity screening and proteomics. **12 Hrs**
- 4. Cell Culture Techniques:** Basic equipments used in cell culture lab. Cell culture media, various types of cell culture, general procedure for cell cultures, isolation of cells, subculture, cryopreservation, characterization of cells and their applications. Principles and applications of cell viability assays (MTT assays). Principles and applications of flow cytometry. **12 Hrs**
- 5. Metabolite Identification:** In-vitro/ in-vivo approaches, protocols and sample preparation. Microsomal approaches (Rat liver microsomes (RLM) and human liver microsomes (HLM) in Met-ID. Regulatory perspectives. In-vitro assay of drug metabolites & drug metabolizing enzymes.  
Drug Product Performance, in vivo: Bioavailability and bioequivalence: Drug product performance, purpose of bioavailability studies, relative and absolute availability. Methods for assessing bioavailability, bioequivalence studies, design and evaluation of bioequivalence studies, study designs, crossover study designs, generic biologics (biosimilar drug products), clinical significance of bioequivalence studies. **12 Hrs**

**References:**

1. Analysis of Drugs in Biological Fluids - Joseph Chamberlain, 2nd Edition. CRC Press, New York. 1995.
2. Principles of Instrumental Analysis - Douglas A Skoog, F. James Holler, Timothy A. Nieman, 5th edition, Eastern press, Bangalore, 1998.
3. Pharmaceutical Analysis - Higuchi, Brochmman and Hassen, 2nd Edition, Wiley – Interscience Publications, 1961.

4. Pharmaceutical Analysis- Modern methods – Part B - J W Munson, Volume 11, Marcel Dekker Series.
5. Practical HPLC Method Development – Snyder, Kirkland, Glaich, 2nd Edition, John Wiley & Sons, New Jercey. USA.
6. Chromatographic Analysis of Pharmaceuticals – John A Adamovics, 2nd Edition, Marcel Dekker, Newyork, USA. 1997.
7. Chromatographic Methods in Clinical Chemistry & Toxicology – Roger L Bertholf, Ruth E Winecker, John Wiley & Sons, New Jercey, USA. 2007.
8. Good Laboratory Practice Regulations, 2nd Edition, Sandy Weinberg Vol. 69, Marcel Dekker Series, 1995.
9. Good Laboratory Practice Regulations – Allen F. Hirsch, Volume 38, Marcel Dekker Series, 1989.
10. ICH, USFDA & CDSCO Guidelines.

### **Teaching-Learning Strategies in brief**

The teaching learning strategies, followed are chalk-board teaching, learning through discussion among the peer group, classroom interaction, quiz, presentations, Q & A session and reflective learning.

### **Assessment methods and weightages in brief**

There are two components of assessment: Internal assessment (25 marks) and End semester examination (75 marks). Internal assessment consists of continuous mode (10 marks) and sessional examinations (15 marks). Continuous mode evaluation is of 10 marks comprising of Attendance -8 marks (calculated as: Percentage of Attendance: Allotment of marks- Less than 80: 0 marks; 80-84: 2 mark; 85-89: 4 marks; 90-94: 6 marks and 95-100: 8 marks), Academic activities

Student teacher interaction-2 marks. There are two Sessional exams (each conducted for 30 marks and computed for 15 marks) and one improvement exam (30 marks and computed for 15 marks). The average marks of two best sessional exams are computed out of 15 marks.

Total Marks are 100 for the subject (Internal Assessment: 25 marks and End Semester Examination: 75 Marks)

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**Name of the Academic Program: M. Pharm (Pharmaceutical Analysis)**

**Course Code: MPA203T**

**Title of the Course: Quality Control and Quality Assurance**

**L-T-P: 4-0-0**

**Credits: 04**

(L=Lecture hours, T=Tutorial hours, P=Practical hours)

### **COURSE OUTCOMES (COs)**

After completing this Course, the students should be able to

**CO-1:** Understand the c-GMP aspects in pharmaceutical industry (**Cognitive level- Understand**).

**CO-2:** Utilize the procedures and instructions to maintain and represent documentation in pharmaceutical industry (**Cognitive level- Apply**).

**CO-3:** Assess quality certifications applicable to pharmaceutical industry (**Cognitive level- Analyse**)

**CO-4:** Utilize the principles and various considerations of Good Laboratory Practices (**Cognitive level- Apply**).

**CO-5:** Identify the responsibilities of QA and QC departments (**Cognitive level- Analyse**).

### **Mapping of Course Outcomes (COs) with Program Outcomes (POs) and Program Specific Outcomes (PSOs)**

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3	PSO4
<b>CO1</b>	3	3	3	3	1	2		1		2	2	3	3	2	1
<b>CO2</b>	3	2	3	3			1		1	3		3	3	2	1
<b>CO3</b>	3	2	2	3	1		1	1		3	1	3	3	3	2
<b>CO4</b>	3	3	3	3	3	2	2	2	2	2	1	3	2	3	3
<b>CO5</b>	3	2	3	3	1	1	2	1	2	3	1	3	2	3	3
<b>CO6</b>	3		3	3		2				1		3		3	1
<b>CO7</b>	3		3	3		2		1		1		3		3	1

Each Course Outcome (CO) may be mapped with one or more Program Outcomes (POs). Write '3' in the box for 'High-level' mapping, 2 for 'Medium-level' mapping, 1 for 'Low'-level' mapping.

### **Detailed Syllabus:**

**Theory**

**60 Hrs**

#### **1. Concept and Evolution of Quality Control and Quality Assurance 12 Hrs**

Good Laboratory Practice, GMP, Overview of ICH Guidelines- QSEM, with special emphasis on Q-series guidelines.

Good Laboratory Practices: Scope of GLP, Definitions, Quality assurance unit, protocol for conduct of non clinical testing, control on animal house, report preparation and documentation.

**2. cGMP guidelines according to schedule M, USFDA (inclusive of CDER and CBER) Pharmaceutical Inspection Convention 12 Hrs**

(PIC), WHO and EMEA covering: Organization and personnel responsibilities, training, hygiene and personal records, drug industry location, design, construction and plant lay out, maintenance, sanitation, environmental control, utilities and maintenance of sterile areas, control of contamination and Good Warehousing Practice. CPCSEA guidelines.

**3. Analysis of raw materials, finished products, packaging materials, in process quality control (IPQC), Developing specification (ICH Q6 and Q3) 12 Hrs**

Purchase specifications and maintenance of stores for raw materials. In process quality control and finished products quality control for following formulation in Pharma industry according to Indian, US and British pharmacopoeias: tablets, capsules, ointments, suppositories, creams, parenterals, ophthalmic and surgical products (How to refer pharmacopoeias), Quality control test for containers, closures and secondary packing materials.

**4. Documentation in pharmaceutical industry: 12 Hrs**

Three tier documentation, Policy, Procedures and Work instructions, and records (Formats), Basic principles- How to maintain, retention and retrieval etc. Standard operating procedures (How to write), Master Formula Record, Batch Formula Record, Quality audit plan and reports. Specification and test procedures, Protocols and reports. Distribution records. Electronic data.

**5. Manufacturing operations and controls: 12 Hrs**

Sanitation of manufacturing premises, mix-ups and cross contamination, processing of intermediates and bulk products, packaging operations, IPQC, release of finished product, process deviations, charge-in of components, time limitations on production, drug product inspection, expiry date calculation, calculation of yields, production record review, change control, sterile products, aseptic process control, packaging.

**Reference Books:**

1. Quality Assurance Guide by organization of Pharmaceutical Procedures of India, 3rd revised edition, Volume I & II, Mumbai, 1996.
2. Good Laboratory Practice Regulations, 2nd Edition, Sandy Weinberg Vol. 69, Marcel Dekker Series, 1995.
3. Quality Assurance of Pharmaceuticals- A compedium of Guide lines and Related materials Vol I & II, 2nd edition, WHO Publications, 1999.
4. How to Practice GMP's – P P Sharma, Vandana Publications, Agra, 1991.
5. The International Pharmacopoeia – vol I, II, III, IV & V - General Methods of Analysis and Quality specification for Pharmaceutical Substances, Excipients and Dosage forms, 3rd edition, WHO, Geneva, 2005.
6. Good laboratory Practice Regulations – Allen F. Hirsch, Volume 38, Marcel Dekker Series, 1989.
7. ICH guidelines
8. ISO 9000 and total quality management

9. The drugs and cosmetics act 1940 – Deshpande, Nilesh Gandhi, 4th edition, Susmit Publishers, 2006.
10. QA Manual – D.H. Shah, 1st edition, Business Horizons, 2000.
11. Good Manufacturing Practices for Pharmaceuticals a plan for total quality control – Sidney H. Willig, Vol. 52, 3rd edition, Marcel Dekker Series.
12. Steinborn L. GMP/ISO Quality Audit Manual for Healthcare Manufacturers and Their Suppliers, Sixth Edition, (Volume 1 - With Checklists and Software Package). Taylor & Francis; 2003.
13. Sarker DK. Quality Systems and Controls for Pharmaceuticals. John Wiley & Sons; 2008.

### **Teaching-Learning Strategies in brief**

The teaching learning strategies, followed are chalk-board teaching, learning through discussion among the peer group, classroom interaction, quiz, presentations, Q & A session and reflective learning.

### **Assessment methods and weightages in brief**

*There are two components of assessment:* Internal assessment (25 marks) and End semester examination (75 marks). Internal assessment consists of continuous mode (10 marks) and sessional examinations (15 marks). Continuous mode evaluation is of 10 marks comprising of Attendance -8 marks (calculated as: Percentage of Attendance: Allotment of marks- Less than 80: 0 marks; 80-84: 2 mark; 85-89: 4 marks; 90-94: 6 marks and 95-100: 8 marks), Academic activities

Student teacher interaction-2 marks. There are two Sessional exams (each conducted for 30 marks and computed for 15 marks) and one improvement exam (30 marks and computed for 15 marks). The average marks of two best sessional exams are computed out of 15 marks.

Total Marks are 100 for the subject (Internal Assessment: 25 marks and End Semester Examination: 75 Marks)

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**Name of the Academic Program: M. Pharm (Pharmaceutical Analysis)**

**Course Code: MPA 204T**

**Title of the Course: Herbal and Cosmetic Analysis**

**L-T-P: 4-0-0 Credits: 4**

(L=Lecture hours, T=Tutorial hours, P=Practical hours)

### **COURSE OUTCOMES (COs)**

After completing this Course, the students should be able to:

**CO 1:** Describe herbal drug standardization as per WHO guidelines. (**Cognitive level- Understand**).

**CO 2:** Compare the natural product monograph from different herbal pharmacopoeias. (**Cognitive level- Analyse**).

**CO 3:** Discuss regulatory requirements for setting up an herbal drug industry. (**Cognitive level- Apply**).

**CO 4:** Apply DNA finger printing technique for the identification of drugs of natural origin (**Cognitive level- Apply**).

**CO 5:** Summarize the types, causes and measures of adulteration. (**Cognitive level- Apply**).

**CO 6:** Determine the Herbal Drug-Drug interaction. (**Cognitive level- Evaluate**).

**CO 7:** Explain the principles of performance evaluation of cosmetic products. (**Cognitive level- Create**).

**CO 8:** Knowledge of various herbal drugs and cosmetics for the treatment of various ailments and diseases. (**Cognitive level- Understand**).

### **Mapping of Course Outcomes (COs) with Program Outcomes (POs) and Program Specific Outcomes (PSOs)**

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3	PSO4
<b>CO1</b>	3	3	3	1	1	1	2	1	1	2	3	1	3	3	2
<b>CO2</b>	3	2	3	1	1	1	2	1	1	2	2	1	3	3	2
<b>CO3</b>	3	2	3	2	1	1	2	1	1	2	2	1	3	3	2
<b>CO4</b>	3	3	3	2	1	1	2	1	1	2	2	1	3	3	2
<b>CO5</b>	3	1	3	3	1	1	2	1	1	2	2	1	3	3	2
<b>CO6</b>	3	3	3	2	1	1	2	1	1	2	2	1	3	3	2
<b>CO7</b>	3	3	3	2	1	1	2	1	1	2	2	1	3	3	2
<b>CO8</b>	3	2	3	2	1	1	2	1	1	2	2	1	3	3	

Each Course Outcome (CO) may be mapped with one or more Program Outcomes (POs). Write '3' in the box for 'High-level' mapping, 2 for 'Medium-level' mapping, 1 for 'Low'-level' mapping.

## Detailed Syllabus

60 Hrs

1. **Herbal remedies- Toxicity and Regulations:** Herbals vs Conventional drugs, Efficacy of herbal medicine products, Validation of Herbal Therapies, Pharmacodynamic and Pharmacokinetic issues. Herbal drug standardization: WHO and AYUSH guidelines.

12 Hrs

2. **Adulteration and Deterioration:** Introduction, types of adulteration/substitution of herbal drugs, Causes and Measure of adulteration, Sampling Procedures, Determination of Foreign Matter, DNA Finger printing techniques in identification of drugs of natural origin, heavy metals, pesticide residues, phototoxin and microbial contamination in herbal formulations.

Regulatory requirements for setting herbal drug industry: Global marketing management, Indian and international patent law as applicable herbal drugs and natural products and its protocol.

12 Hrs

3. **Testing of natural products and drugs:** Effect of herbal medicine on clinical laboratory testing, Adulterant Screening using modern analytical instruments, Regulation and dispensing of herbal drugs, Stability testing of natural products, protocol.

**Monographs of Herbal drugs:** Study of monographs of herbal drugs and comparative study in IP, USP, Ayurvedic Pharmacopoeia, American herbal Pharmacopoeia, British herbal Pharmacopoeia, Siddha and Unani Pharmacopoeia, WHO guidelines in quality assessment of herbal drugs.

12 Hrs

4. **Herbal drug-drug interaction:** WHO and AYUSH guidelines for safety monitoring of natural medicine, Spontaneous reporting schemes for bio drug adverse reactions, bio drug-drug and bio drug-food interactions with suitable examples. Challenges in monitoring the safety of herbal medicines.

12 Hrs

5. **Evaluation of cosmetic products:** Determination of acid value, ester value, saponification value, iodine value, peroxide value, rancidity, moisture, ash, volatile matter, heavy metals, fineness of powder, density, viscosity of cosmetic raw materials and finished products. Study of quality of raw materials and general methods of analysis of raw material used in cosmetic manufacture as per BIS.

Indian Standard specification laid down for sampling and testing of various cosmetics in finished forms such as baby care products, skin care products, dental products, personal hygiene preparations, lips sticks. Hair products and skin creams by the Bureau Indian Standard.

## References

1. Pharmacognosy by Trease and Evans
2. Pharmacognosy by Kokate, Purohit and Gokhale



3. Quality Control Methods for Medicinal Plant, WHO, Geneva
4. Pharmacognosy & Pharmacobiotechnology by Ashutosh Kar
5. Essential of Pharmacognosy by Dr. S. H. Ansari
6. Cosmetics – Formulation, Manufacturing and Quality Control, P.P. Sharma, 4th edition, Vandana Publications Pvt. Ltd., Delhi
7. Indian Standard specification, for raw materials, BIS, New Delhi.
  - a. Indian Standard specification for 28 finished cosmetics BIS, New Delhi
  - b. Harry's Cosmeticology 8th edition
  - c. Suppliers catalogue on specialized cosmetic excipients
  - d. Wilkinson, Moore, seventh edition, George Godwin. Poucher's Perfumes, Cosmetics and Soaps
8. Hilda Butler, 10th Edition, Kluwer Academic Publishers. Handbook of Cosmetic Science and Technology, 3rd Edition

### **Teaching-Learning Strategies in brief**

The teaching learning strategies, followed are chalk-board teaching, learning through discussion among the peer group, classroom interaction, quiz, presentations, Q & A session and reflective learning.

### **Assessment methods and weightages in brief**

There are two components of assessment: Internal assessment (25 marks) and End semester examination (75 marks). Internal assessment consists of continuous mode (10 marks) and sessional examinations (15 marks). Continuous mode evaluation is of 10 marks comprising of Attendance -8 marks (calculated as: Percentage of Attendance: Allotment of marks- Less than 80: 0 marks; 80-84: 2 mark; 85-89: 4 marks; 90-94: 6 marks and 95-100: 8 marks), Academic activities

Student teacher interaction-2 marks. There are two Sessional exams (each conducted for 30 marks and computed for 15 marks) and one improvement exam (30 marks and computed for 15 marks). The average marks of two best sessional exams are computed out of 15 marks.

Total Marks are 100 for the subject (Internal Assessment: 25 marks and End Semester Examination: 75 Marks)

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**Name of the Academic Program: M. Pharm-II Semester (Pharmaceutical Analysis)**

**Course Code: MPA 205T**

**Title of the Course: Pharmaceutical Analytical Techniques Practical's -II**

**L-T-P: 0-0-12**

**Credits: 6**

(L=Lecture hours, T=Tutorial hours, P=Practical hours)

### **COURSE OUTCOMES (COs)**

After completing this Course, the students should be able to:

**CO-1:** Acquire the cognitive, technical and creative skills and apply established knowledge and practice concerning modern analytical hyphenated instrumentation and measurement techniques to a range of situations (**Cognitive Level: Understand and Apply**)

**CO-2:** Investigate and solve qualitative and quantitative problems in the analytical / pharmaceutical sciences, both individually and in teams (**Cognitive Level: Apply**)

**CO-3:** Interpret the data obtained in various spectroscopic methods (**Cognitive Level: Analyse**)

**CO-4:** Describe the instrumentation required for the various separation techniques and their associated operating principles (**Cognitive Level: Understand**)

**CO-5:** Understand the principle and theory involved in the various instrumental techniques (**Cognitive Level: Understand**)

**CO-6:** Exhibit the knowledge of analysing the drugs in biological matrices and bio analytical method validation (**Cognitive Level: Understand**)

### **Mapping of Course Outcomes (COs) with Program Outcomes (POs) and Program Specific Outcomes (PSOs)**

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
<b>CO1</b>	3			2					3			3		
<b>CO2</b>	3										3	3		
<b>CO3</b>	3	2	3	2					1	2	3	3		
<b>CO4</b>	3		3							2	3	3		
<b>CO5</b>	3	3	2	3					2	2	3	3		
<b>CO6</b>	3	2									3			

Each Course Outcome (CO) may be mapped with one or more Program Outcomes (POs). Write '3' in the box for 'High-level' mapping, 2 for 'Medium-level' mapping, 1 for 'Low'-level' mapping.

## Detailed Syllabus

12 Hrs/Week

1. Comparison of absorption spectra by UV and Wood ward – Fiesure rule.
2. Interpretation of organic compounds by FT-IR.
3. Interpretation of organic compounds by NMR.
4. Interpretation of organic compounds by MS.
5. Determination of purity by DSC in pharmaceuticals.
6. Identification of organic compounds using FT-IR, NMR, <sup>13</sup>CNMR and Mass spectra.
7. Bio molecules separation utilizing various sample preparation techniques and Quantitative analysis of components by gel electrophoresis.
8. Biomolecules separation utilizing various sample preparation techniques and Quantitative analysis of components by HPLC techniques.
9. Isolation of analgesics from biological fluids (Blood serum and urine).
10. Protocol preparation and performance of analytical/Bioanalytical method validation.
11. Protocol preparation for the conduct of BA/BE studies according to guidelines.
12. Quantitative analysis of rancidity in lipsticks and hair oil.
13. Determination of aryl amine content and Developer in hair dye.
14. Determination of foam height and SLS content of Shampoo.
15. Determination of total fatty matter in creams (Soap, skin and hair creams).
16. Determination of acid value and saponification value.
17. Determination of calcium thioglycolate in depilatories.
18. Determination of total reducing sugar.
19. Determination of proteins.
20. Determination of saponification value, Iodine value, Peroxide value, Acid value in food products.
21. Determination of fat content and rancidity in food products.
22. Analysis of natural and synthetic colors in food.
23. Determination of preservatives in food.
24. Determination of pesticide residue in food products.
25. Analysis of vitamin content in food products.
26. Determination of density and specific gravity of foods.
27. Determination of food additives.

## Reference Books

1. Spectrometric Identification of Organic compounds - Robert M Silverstein, Sixth edition, John Wiley & Sons, 2004.
2. Principles of Instrumental Analysis - Douglas A Skoog, F. James Holler, Timothy A. Nieman, 5th edition, Eastern press, Bangalore, 1998.
3. Instrumental methods of analysis – Willards, 7th edition, CBS publishers.
4. Organic Spectroscopy - William Kemp, 3rd edition, ELBS, 1991.
5. Quantitative analysis of pharmaceutical formulations by HPTLC - P D Sethi, CBS Publishers, New Delhi.
- 6.

7. Analysis of drugs in biological fluids - Joseph Chamberlain, 2nd Edition. CRC Press, Newyork. 1995.
8. Principles of Instrumental Analysis - Doglas A Skoog, F. James Holler, Timothy A. Nieman, 5th edition, Eastern press, Bangalore, 1998.
9. Pharmaceutical Analysis - Higuchi, Brochmman and Hassen, 2nd Edition, Wiley – Interscience Publications, 1961.
10. Pharmaceutical Analysis- Modern methods – Part B - J W Munson, Volume 11, Marcel Dekker Series

### **Teaching-Learning Strategies in brief**

The teaching learning strategies, followed are chalk-board teaching, learning through discussion among the peer group, classroom interaction, quiz, presentations, Q & A session and reflective learning.

### **Assessment methods and weightages in brief**

#### **Practical**

There are two components of assessment: Internal assessment (50 marks) and End semester examination (100 marks). Internal assessment consists of continuous mode (20 marks) and sessional examinations (30 marks). Continuous mode evaluation is of 10 marks comprising of Attendance- 10 marks (calculated as: Percentage of Attendance: Allotment of marks- Less than 80: 0 marks; 80-84: 2.5 mark; 85-89:5 mark; 90-94: 7.5 marks and 95-100: 10 marks) and based on practical records, regular viva voce, etc. -10 marks. There are two Sessional exams (each conducted for 40 marks and computed for 30 marks) and one improvement exam (40 marks and computed for 30 marks). The average marks of two best sessional exams are computed out of 30 marks.

Total Marks are 150 for the subject (Internal Assessment: 50 marks and End Semester Examination: 100 Marks)

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